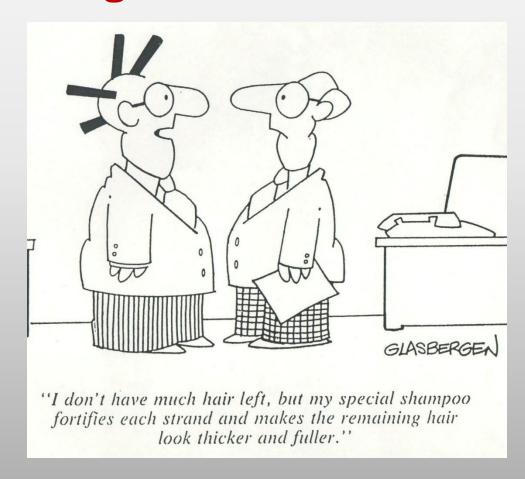
## Visual Aids Make Chemistry Fun

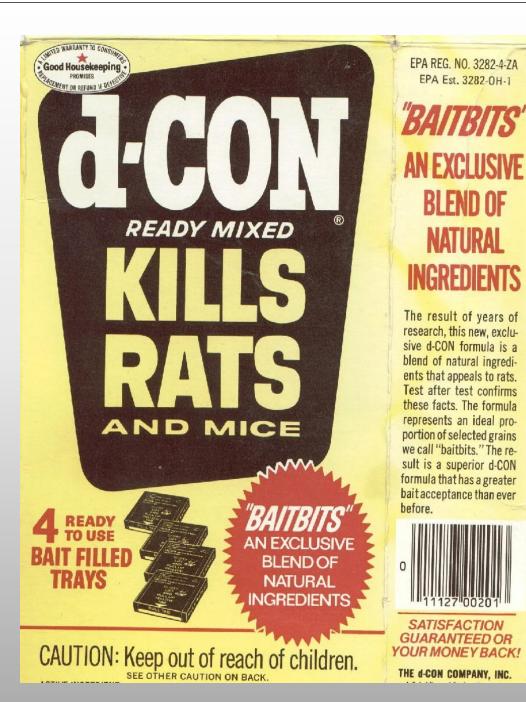
John Halal
www.chemistrysimplified.com
john@chemistrysimplified.com

#### **Effective Visual Aids**

- Create Interest and add Excitement
- Reinforce Essential Ideas
- Unforgettable Learning Experience

"I don't have much hair left, but my special shampoo fortifies each strand and makes the remaining hair look thicker and fuller."





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## **Natural and Organic**

- Poison Ivy
- Natural Gas
- Carbon Monoxide
- Cow Manure
- Motor Oil

## Inorganic

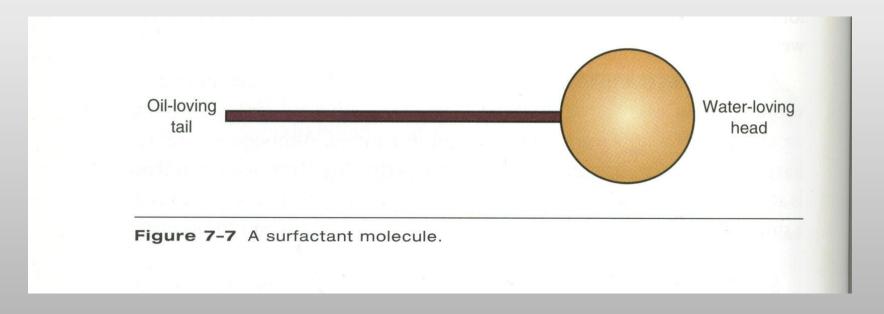
- Water H2O
- Oxygen O2

## Synthetic or Natural

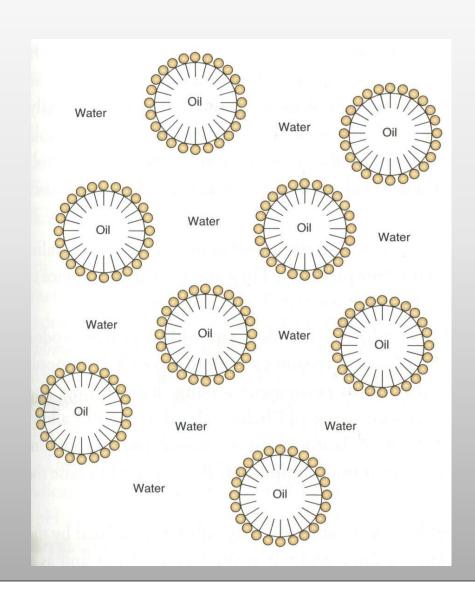


#### **Emulsions**

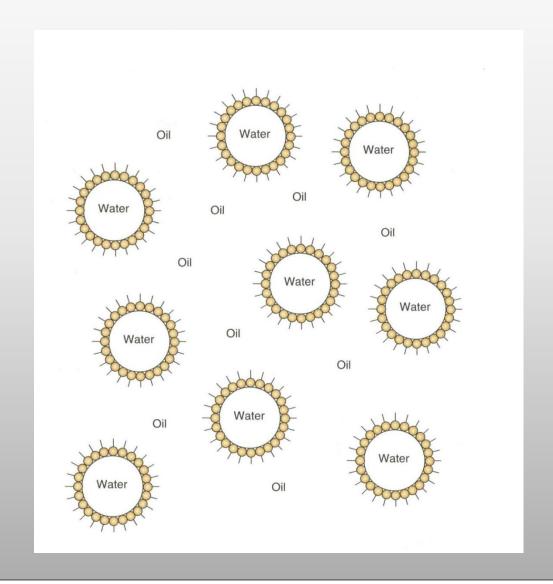
- Oil & Water Do Not Mix
- Surfactant (Surface Active Agent)
- Surfactants Emulsify Oil & Water



#### Oil in Water Emulsion

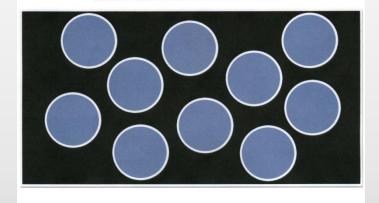


#### Water in Oil Emulsion

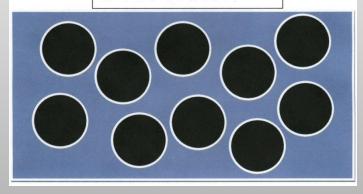


#### Polka Dots

Water in Oil



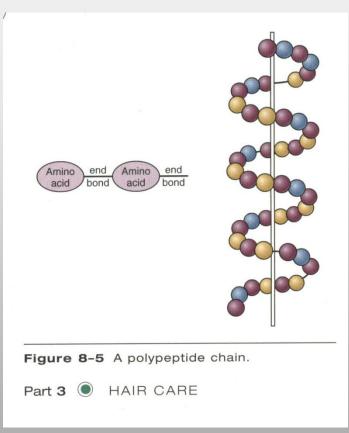
Oil in Water

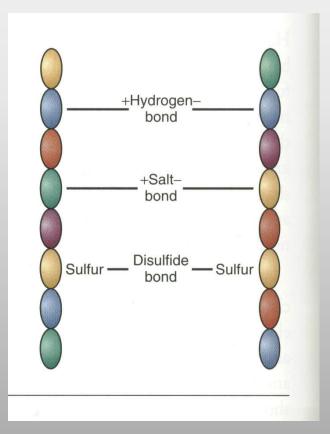


#### Hair Structure

- End Bonds Polypeptide Bonds
- Side Bonds

Wet Sets, Thermal Styling, Perms & Texturizers





#### **Thio Permanent Waving**

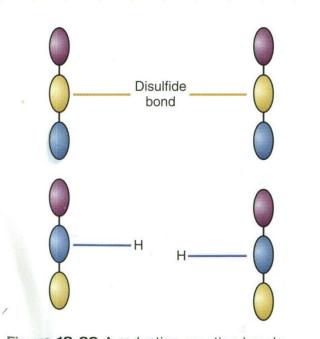
Part 1, Thio Breaks Side Bonds
 Wrapping Creates Physical Tension

• Part 2, Neutralizer Reforms Side Bonds In New Pairs

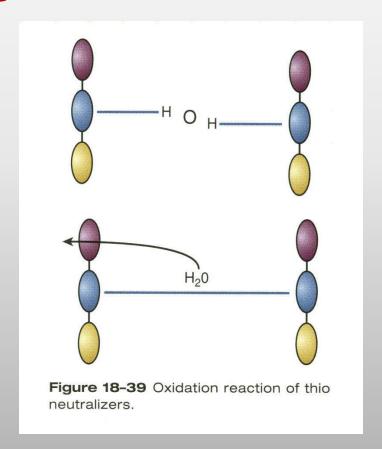
13



#### Thioglycolate/Thio Processing is Reduction Neutralizing is Oxidation



**Figure 18-32** A reduction reaction breaks disulfide bonds during the permanent waving process.



- Disulfide Bonds Broken
  - 1. Thio
  - 2. Extreme heat

Hair Dryer – Hot Irons – Boiling Water

 Disulfide Bonds Reformed When Hair is Dried Air Oxidation (Neutralization) 72 hours – 3 Days

#### Student Project

Wrap a swatch of hair on a Perm Rod.

Apply Perm Solution.

Wait 3 Days

OR

Wrap a swatch of hair on a Perm Rod

Place in Boiling Water for 20 minutes.

Remove from water.

Wait 3 days.

# Alkaline/Acid/Exothermic Permanent Waves

	Alkaline	Acid	Exothermic
Waving Lotion	Ammonium Thioglycolate	Ammonium Thioglycolate	Ammonium Thioglycolate
Activator	None	Glyceryl Thioglycolate	Hydrogen Peroxide
Neutralizer	Hydrogen Peroxide	Hydrogen Peroxide	Hydrogen Peroxide

#### Hydroxide Relaxers

- Hydroxide relaxers break disulfide bonds by lanthionization.
- Rinsing converts disulfide bonds into lanthionine bonds.
- Hydroxide neutralizers are not oxidizers
- Hydroxide neutralizers do not rebuild disulfide bonds.
- Hydroxide neutralizers neutralize alkalinity.
- Hair Treated with Hydroxide cannot be curled with Thio.

# The Natural Ionization of Water $H_2O \rightarrow H^+ + OH^-$

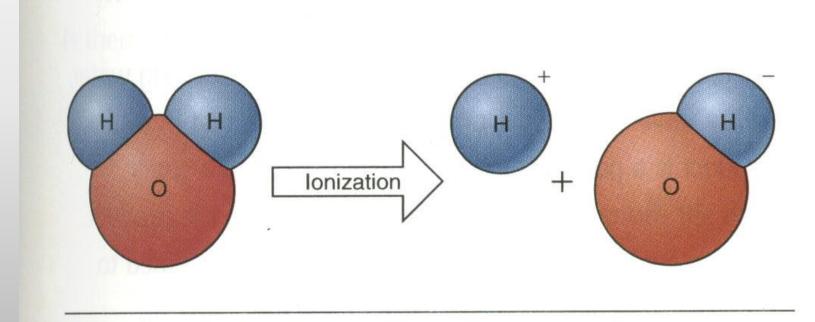


Figure 7-10 The ionization of water.

#### The Ionization of Water Means

- pH is only possible because of the ionization of water. Only aqueous (water) solutions have a ph. Oil and alcohol do not have a pH.
- Pure water isn't neutral because it is neither acidic nor alkaline. Pure water is neutral because it is an equal balance of both acid and alkali.

# The pH Scale

	H <sup>+</sup> Hydrogen Ion		OH <sup>-</sup> Hydroxide Ion	
рН	Exponential Notation	With Decimal	Exponential Notation	With Decimal
0	1 × 10 <sup>-0</sup>	1.	$1 \times 10^{-14}$	.00000000000001
1	$1 \times 10^{-1}$	.1	$1 \times 10^{-13}$	.000000000001
2	$1 \times 10^{-2}$	.01	$1 \times 10^{-12}$	.00000000001
3	$1 \times 10^{-3}$	.001	$1 \times 10^{-11}$	.0000000001
4	$1 \times 10^{-4}$	.0001	$1 \times 10^{-10}$	.000000001
5	$1 \times 10^{-5}$	.00001	$1 \times 10^{-9}$	.00000001
6	$1 \times 10^{-6}$	.000001	$1 \times 10^{-8}$	.0000001
7	$1 \times 10^{-7}$	.0000001	$1 \times 10^{-7}$	.0000001
8	$1 \times 10^{-8}$	.0000001	$1 \times 10^{-6}$	.000001
9	$1 \times 10^{-9}$	.00000001	$1 \times 10^{-5}$	.00001
10	$1 \times 10^{-10}$	.000000001	$1 \times 10^{-4}$	.0001
11	$1 \times 10^{-11}$	.0000000001	$1 \times 10^{-3}$	.001
12	$1 \times 10^{-12}$	.00000000001	$1 \times 10^{-2}$	.01
13	$1 \times 10^{-13}$	.000000000001	$1 \times 10^{-1}$	.1
14	$1 \times 10^{-14}$	.00000000000001	$1 \times 10^{-0}$	1.

Figure 9-3 The quantities of the pH scale expressed as pH, exponential notation, and with a decimal point.

#### Hydroxide Relaxers

- Lye No-Mix Sodium Hydroxide NaOH
- No Lye No-Mix Potassium Hydroxide KOH
- No-Lye No-Mix Lithium Hydroxide LiOH
- No-Lye Mix 2 Parts Guanidine Hydroxide GOH

# Hydroxide Neutralizers Acid – Alkali Neutralization H<sup>+</sup> + OH<sup>-</sup> → H<sub>2</sub>O

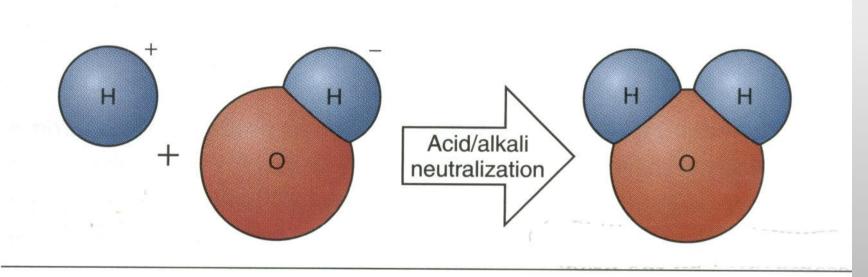


Figure 7-12 Acid/alkali neutralization reaction.

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#### Hydroxide Neutralizers

- Rinsing converts the disulfide bonds to lanthionine bonds.
- Hydroxide neutralizers are not oxidizers.
- Hydroxide neutralizers do not rebuild disulfide bonds.
- Hydroxide neutralizers neutralize alkalinity.
- Hair Treated with Hydroxide cannot be curled with Thio.



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